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**Contribution à l'étude des
sulfures de cuivre non stoechiométriques**

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The anodic oxidation of cuprous sulphide has been studied using a carbon paste electrode. Typical voltammetric curves are obtained. At slow rates, the oxidation of Cu_2S in CuS and Cu^{2+} occurs via the formation of non stoichiometric compounds: $\text{Cu}_{1,92}\text{S}$, $\text{Cu}_{1,77}\text{S}$, $\text{Cu}_{1,60}\text{S}$ and $\text{Cu}_{1,31}\text{S}$.

Actually I evidenced

"the formation of non stoichiometric compounds: $\text{Cu}_{1,92}\text{S}$, $\text{Cu}_{1,77}\text{S}$, $\text{Cu}_{1,60}\text{S}$ and $\text{Cu}_{1,31}\text{S}$ "

during my DEA (=Master thesis), and on this subject this is my work...

Pierre Vitorge